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STABILITY INDICATIµG ASSAY METHOD DEVELOPMENT AND VALIDATION FOR ONDANSETRON HYDROCHLORIDE AND PANTOPRAZOLE SODIUM IN BULK AND PHARMACEUTICAL DOSAGE FORM

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Nitya Patel Department of Quality Assurance, Radharaman Institute of Pharmaceutical Sciences, Bhopal, Madhya Pradesh, India ⊠ nityapatelpcp@gmail.com Keywords Pantoprazole, RP-HPLC, Ondansetron, Antiemetic. Received 15/07/2019 Reviewed 17/07/2019 Revised/ Accepted 21/07/2019

ABSTRACT

RP-HPLC method showed adequate linearity from 100-600 μ g/mL for PAN (Pantoprazole sodium) and 10- 60 μ g/ml for OND (Ondansetron). The mean recoveries for all methods were found in between 98 %-102 % for both the drugs. The RP-HPLC method successfully separated Pantoprazole and Ondansetron from degradation products formed under stress conditions like acidic, alkali, oxidative, photolytic and thermal. PAN degraded significantly under acidic, neutral, oxidative, photolytic and thermal conditions and gave 1 degradation products respectively, whereas OND degraded significantly under oxidative conditions and gave 1 degradation product each condition. Both the drugs were found to be stable under alkali conditions. HPTLC method showed adequate linearity from 200-1200 μ g/band for OND and 2000-12000 μ g/band for PAN.

INTRODUCTION

The RP-HPLC method developed for quantitative and qualitative analysis of pantoprazole and Ondansetron tablets was rapid, simple, accurate, precise and specific. Here conclude that this method is one of best method available in market because it is Economically proved. Recovery study on tablet formulation gave accurate prediction of good relation between label claim and practical value. Method is specific due to it shows no interference in analyte peak of any unknown or known compounds. Method was validated as per USP acceptance criteria and ICH guidelines. Hence present validated method can be confidently utilize as an alternative method to reporting of result in routine practice or release of product report in quality control compliance (1).

Ondansetron hydrochloride is the prototype of new class of antiemetic drugs used for the prevention of nausea and vomiting associated with highly emetogenic cancer chemotherapy, radiotherapy or anesthesia and surgery.

Ondansetron is serotonin 5-HT3 receptor antagonist used in mainly as an antiemetic. Its effects are through to be on both peripheral and central nerves. It reduce the activity of the vagus nerve which activates the vomiting center in medulla oblongata, and also block serotonin receptor in chemotherapy trigger zone.

Pantoprazole sodium is proton pump inhibitor. It is used in treatment of gastro esophageal reflux disease, gastric ulcer and duodenal ulcer. It accumulates in the acidic compartment of parietal cells and is converted to the active form, which binds to hydrogen-potassium- ATPase at the secretory surface of gastric parietal cells. Inhibition of hydrogen- potassium- ATPase blocks the final step of gastric acid production, leading to inhibition of both basal and stimulated acid secretion.

MATERIAL AND METHOD

In RP-HPLC method, chromatographic separation was achieved on Phenomenex, C_8 (250× 4.6) mm 5µm column using Methanol: ACN: Water (20:30:50%) as the mobile phase with detection at 216 nm. Both the drugs were subjected to acid, alkali, oxidative, thermal and photolytic conditions individually and in stress combination whereas tablet formulation was subjected to thermal and photolytic stress conditions. Both the methods were validated as per ICH guidelines. In HPTLC method, optimized on TLC plate pre-coated with silica gel 60F 254 using Dichloromethane: Methanol (9.0:0.7) as

mobile phase and scanning the plate at 290 nm.

DETERMINATION OF λ max



Mode: Spectrum Scan, Speed: Medium Wavelength range: 400-200 nm Absorbance scale: 0.00A - 2.00A Initial base line correction: Methanol Concentration: 10 ppm of OND & 10ppm PAN

CHROMATOGRAPHIC CONDITIONS

Column- Phenomenex Luna C₈ (250 mm \times 4.6 mm, 5 μ m

HPLC system: LC 2010CHT, Shimadzu

UV detector

Mobile phase: Methanol : Acetonitrile:

Water (20:30:50 % v/v/v)

Flow rate: 1.0 mL/min

Detection wavelength: 216 nm

Column Temperature: 40°C

Total run time: 20 min

Injection Volume: 20 µL

Diluent: All the final dilution of sample and standard done with Mobile Phase

VALIDATION OF DEVELOPED HPLC METHOD AND DISCUSSION

LINEARITY AND RANGE

The linearity of PAN and OND were found between 100-600 μ g/mL and 10-60 μ g/mL respectively. The calibration data is presented in Table and correlation coefficient and regression line equation analysis presented in Figure

Concentration		Mean Peak Area*		% RSD	
(µg/mL)		± S.D			
PAN	OND	PAN	OND	PAN	OND
100	10	1639082	147182	0.56	0.17
		5 ±	5±526		
		91490	4		
200	20	3248762	302548	0.31	0.20
		$1 \pm$	2±589		
		100265	7		
.300	30	5215489	458795	0.24	0.22
		6 ±	4±998		
		122569	5		
400	40	7012458	584875	0.29	0.34
		9 ±	0±197		
		201548	45		
500	50	8241258	735483	0.43	0.27
		6 ±	2±201		
		358745	59		
600	60	1025482	869874	0.45	0.34
		$1 \pm$	5±298		
		458562	54		

Calibration Curve of PAN



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Calibration Curve of OND

LOD AND LOQ

I	PAN	OND		
LOD µg/mL	LOQ µg/mL	LOD LOQ µg/mL µg/mL		
17.65	53.50	1.21	3.67	

The Limit of detection (LOD) was found to be 17.65 μ g/mL and 0.03 μ g/mL; while the Limit of quantification (LOQ) was found to be 5.48 μ g/mL and 0.09 μ g/mL for PAN and OND, respectively.

FORCED DEGRADATION STUDY

% Degradation of PAN and OND in various stress condition are shown in table. From degradation study it was found that PAN marginally degraded in acidic. was oxidative, photolytic and thermal condition and stable in alkali conditions whereas OND was significantly degraded in oxidative conditions and marginally stable in acidic, alkali, photolytic and thermal and conditions.

Sr	Stress Type	Stress Conditions
1	Acid hydrolysis	0.1 N HCl at RT for 5 hrs
2	Alkali hydrolysis	0.1 N NaOH at RT for 5 hrs
3	Neutral hydrolysis	Water at RT for 24 hrs
4	Oxidative Degradation	3 % H O at RT for 5 hrs
4	Photolytic Degradation	UV 254 nm for 8 hrs
5	Thermal Degradation	At 60 for 48 hrs

Acid hydrolysis

Degrade about 19.2 %. Alkali hydrolysis No significant degradation observed. Neutral hydrolysis Degrade about 6.01 %. Oxidative Degradation Degrade about 22 %. Photolytic Degradation Degrade about 91.3 %. Thermal Degradation Degrade about 19.96. 533 PRECISION 1

Repeatability (Intra-day Precision):

Reproducibility was determined by analyzing PAN and OND standard solution in the range of 200, 400, and 600 μ g/mL of PAN and 20, 40 and 60 μ g/mL of OND in different laboratories. Calculate % RSD for PAN and OND.

Conclusion

The % RSD for Intra-day precision was found to be 0.18-0.63 % for PAN and 0.12- 0.16 % for OND.

Inter-day Precision

Inter-day precision was determined by analyzing of PAN and OND standard solutions in the range 200, 400, and 600 μ g/mL of PAN and 20, 40 and 60 μ g/mL of OND in triplicate in different days. Calculate % RSD for PAN and OND.

Conclusion

The % RSD for Inter-day precision was found to be 0.11-0.80 % for PAN and 0.17-0.43% for OND.

SPECIFICITY

Specificity is the term extent that analyte may be exist without interference from other related compound in a mixture. Specificity able to differentiate all possible impurities by applying forced stress testing. When in any chromatogram analyte peak not affected from other known or unknown impurity it indicates that chromatographic parameters good.

- a) There should not be any interference from blank, excipient and reagent peaks with main peak.
- b) The peak purity index for the main peaks and degradation product peaks in standard preparation and sample preparation should be equa to or more than 0.990.

ACCURACY AS RECOVERY

Accuracy of the method was confirmed by recovery study from marketed formulation at three level of standard addition. Percentage recovery was found to be in range, for PAN 99-101.1 % and for OND 99-101.41 %.

	Amount of Test	Amount of Std	Mean Peak	Amount found	% Recovery
Level	Solution	added	area	(μg/mL)	± RSD
	(µg/mL)	(µg/mL)	± SD		
	% Re	ecovery data	u of PAN		
80%	100	80	3045854	178	99.11±0.4
			5±12356		0
100%	100	100	3356897	196.6	99±0.59
			4±19874		
			5		
120%	100	120	3801456	222	101.1±0.6
			1±38014		8
			56		
	% F	Recovery da	ta of OND		
80 %	10	8	26403	18.2	101.41±
			85±54		0.20
			66		
100 %	10	10	28554	19.7	99±0.28
			22 ± 80		
			24		
120 %	10	12	32120	22.2	101.0±0.
			64±10		31
			125		

ANALYSIS OF MARKETED FORMULATION

Applicability of proposed method was tested by analysing the commercially available Vomizen P tablet. The results are shown in table

VomizenP	Amount	%Assay	% RSD
	of drug	(n=6)	
	(mg/tablet)		
PAN	40	101.01	0.13
OND	4	101.3	0.17

Analysis of market formulation

Preparation of standard stock solution of PAN

Accurately weighed quantity of 10 mg of PAN was transferred into 10 mL volumetric flask, dissolved and diluted up to mark with methanol. This was a stock solution having strength of 1000 μ g/mL of PAN. From this solution, 5 mL of solution was pipetted out and diluted up to 10 mL to get 500 μ g/mL of PAN

Preparation of standard stock solution of OND

Accurately weighed quantity of 1mg of OND was transferred into 10 mL volumetric flask, dissolved and diluted up to mark with methanol.

This was a stock solution having strength of 100 μ g/mL of OND. From this solution, 5 mL of solution was pipetted out and diluted up to 10 mL to get 50 μ g/mL of OND.

Preparation of stock solution of synthetic mixture of PAN and OND

Accurately weighed quantity of 10 mg of PAN and 1 mg of OND was transferred into 10 mL volumetric flask, dissolved and diluted up to mark with methanol. This was a stock solution having strength of 1000 μ g/mL of PAN and 100 μ g/mL of OND. From this solution, 5mL of solution was pipetted out and diluted up to 10 mL to get 500 μ g/mL of PAN and 50 μ g/mL of OND.

Test solution of Pantoprazole and Ondansetron

10 tablets were weighed and powdered; a quantity of tablet powder equivalent of 40 mg PAN and 4 mg of OND was weighed accurately and transferred to a 100 mL volumetric flask. The tablet powder was dissolved in methanol with aid of ultra- sonication, diluted up to mark with same and filtered through a whatman filter paper

CONCLUSION

The suitable chromatographic methods (RP-HPLC, HPTLC) were developed and validated for estimating OND and PAN in tablet dosage form. HPLC method was stability indicating as it achieved separation of both drugs from potential degraded products. More degradation has been observed for either of drugs in combination than degradation of such single drug

REFERENCE

- Rang HP, Dale MM, Ritter JM, 1999. In Pharmacology; 4th Edition, Churchill Livingston, New York, pp 366, 367&363.
- Tripathi KD, 2001. In Pharmacology, Essentials of medical pharmacology, 6th Edition, Jaypee publication, New Delhi, pp 163,169,633,646.
- Sharma PP, 2001. How to Practice GMPs and Good Manufacturing Practices; 3rd Edn; Vandana Publication Pvt Ltd, pp 214.
- Beckett AH, Stenlake JB, 1997. Practical pharmaceutical chemistry; 4th Edn; CBS Publishers and Distributors, New Delhi, pp 293-304.
- Ewing GW, 1985. Instrumental Methods of Chemical Analysis; 5th Edn; McGraw-Hill Book Company, New York, pp 1-7.
- Kasture AV, Mahadik KR, Wadodker SG, More HN, 2006. Instrumental methods of Pharmaceutical analysis; 14th Edn; Nirali Prakashan,, pp 1-30.
- Mendham J, Denney RC, Barnes JD, Thomas MJ, 2004. Vogel's Text Book of Quantitative Analysis; 6th Edn; Pearson Education Limited, pp 1-10.

- Snyder LR, Kirkland JL, Glajch JL, 1977. Practical HPLC Method Development; 2nd Edn; Wiley Inter science, New York, pp 1-26.
- Sethi PD, 1996. HPTLC: Quantitative analysis of Pharmaceutical formulation; 1st Edn; CBS publisher, Delhi, pp 3-23
- Kulkarni GT, Gowthamarajan K, Suresh B, 2004. "Stability Testing of Pharmaceutical Products: An Overview", *Indian J Pharm. Edu. Res*, 38(4), pp 194.
- Achrya MM, 1999. "Pharmaceuticals Stability Testing and Studies: An Overview", *the Eastern Pharmacist*, 42(497), pp 31-33.
- Singh SS, Bakshi M, 2002.
 "Development of Validated Stability Indicating Assay Methods: Critical Review", J. Pharm. Biomed Anal., 28, pp 1011-1040.
- 13. WHO TRS, 2006. 937,40th report, Appendix4, Analytical Method Validation, pp 136-140.
- 14. Ravichandran Vshalini S, Sundram KM, Harish R, 2010. "Validation of analytical methods – strategies & importance", *Int. J. Pharm. Pharm. Sci.*, 2(3), pp 18-22
- 15. International Conference on Harmonization (ICH) of Technical Requirements for the Registration of

Pharmaceuticals for Human Use, Validation of analytical procedures: Methodology, ICH-Q2B, Geneva, pp. 1-17.

- Lippincott's, 2004. Illustrated Reviews: Pharmacology; 2nd Edition; pp 298,307-308.
- Indian Pharmacopoeia, 2010. 6th Edition, vol.-III, Government of India, Ministry of Health & Family Welfare, Indian

Pharmacopoeia commission, pp 1815, 1816, 1819, 1820.

 Merck Index, 2013. Drugs And Biological; 15th Edition; RSC Publishing Thomas Graham House, Cambridge, UK, pp 6942.