



Research article

Design, synthesis and antiinflammatory activity of substituted chromones

Vishal Gupta¹, S K Yadav*²¹ Millennium College of Pharmacy, Bhopal, Madhya Pradesh, India² Ravishankar College of Pharmacy, Bhopal, Madhya Pradesh, India

Corresponding author: S K Yadav ✉ vishalpharmacy@rediffmail.com,

Ravishankar College of Pharmacy, Bhopal, Madhya Pradesh, India

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ABSTRACT

The Chromones and other related ring systems have several interesting biological activities such as anti-inflammatory effects. Chromones also having anti-bacterial and antifungal activities from heterocyclic substituent's at 2- position and exhibit good phosphodiesterase-IV inhibition activity and some chromones have potential HIV- inhibition activity as well as Spiro amine derivatives having anti-arrhythmic activity. performed reaction having catalyzed Claisen rearrangement SnCl₄ of allyl aryl ethers followed by cyclization gives oxygen heterocycles. By utilizing SnCl₄ as lewis acid the rearrangement reaction is affected under mild condition. The anti-inflammatory activity of the synthesized compounds was determined by rat paw edema method. Some of the compounds were found to be potent.

Keywords: Chromones, Inflammation, Melting points, NMR, IR.

INTRODUCTION

Non-steroidal anti-inflammatory drugs (NSAID's) are widely used to treat pain, fever and inflammatory condition including osteoarthritis. Several research studies exhibits the antiinflammatory properties of chromones. Chromones are also reported as anti-microbial anti-cancer anti-allergic and coronary vasodilator. Claisen rearrangement is generally affected by prolong heating of an aryl allyl ether or by heating in vacuum at high temperature. Lewis acids are known to cause the Claisen rearrangement under mild conditions. The use of SnCl₄ as lewis acid act as a catalyst for Claisen rearrangement is not reported. We propose the SnCl₄ catalysed Claisen rearrangement of 7-allyloxy chromones to afford 7 hydroxy 6/8 allylchromones and 2'-methyl-2', 3'-dihydrofuronochromone ^[1].

7- Hydroxyl chromones(2a-d) and 7-allyloxychromones(3a-d) were synthesized as per the reported methods(9,10).

7hydroxy-6/8-allylchromone (4a-d) and 2'-methyl-2',3'-dihydrofuranochromones (5a-d) were synthesized by the treatment of 7-allyloxychromones(3a-d) with SnCl₄ in dry THF (Scheme-1) ^[2].

MATERIAL AND METHOD

Synthesis of 7- Hydroxyl chromones (2a-d) and 7-allyloxychromones (3a-d)

10 mmol's of 1(a-d) and acetic anhydride were refluxed with 5% hydrochloric acid, methanol and sodium acetate at 180°C for 6h. The solution was filtered and the products (2a-d) were recovered. 10mmols of 2(a-d) and 20ml of propylene bromide were refluxed with acetone and potassium carbonate for 5h to obtain compounds 3(a-d).

Synthesis of 7-hydroxy-6/8-allylchromone (4a-d) and 2'-methyl-2', 3'-dihydrofuranochromones (5a-d):

10 mmol's of 3(a-d) and SnCl₄ in dry THF (10ml) were stirred for 14 h under N₂ atmosphere at 0-5°C. The solution was filtered and solvent is removed under reduced pressure. The resulting mixture was chromatographed over silica gel. Elution with benzene-ethyl acetate (1:1, v/v) mixture gave two products 4 and 5. These compounds were crystallized from benzene. The compounds were subjected for NMR studies. **4a** δ 1.95 (s, CH₃₋₃), 2.35(s, CH₃₋₂), 7.65(ar, H-5, J=10H-2), 3.52(m, H-1'), 5.05(dd, H-3', J=10, 1H-2), 9.95(dd, H-3', J=16, 1H-2), 5.90(m-H-2'), 10.20(s, 7-OH). **4b** δ 1.96 (s, CH₃₋₃), 2.38(s, CH₃₋₂), 7.62(s, H-5), 3.59(m, H-1'), 5.91(m-H-2), 5.10(dd, H-3', J=10, 1H-2), 4.99(dd, H-3', J=16, 1H-2). **4c**. δ 1.97 (s, CH₃₋₃), 2.39(s, CH₃₋₂), 7.64(s, H-5), 3.49(m, H-1'), 5.90(m-H-1'), 5.12(dd, H-3', J=10, 1H-2), 4.99(dd, H-3', J=16, 1H-2). **4d** δ 2.10 (s, CH₃₋₃), 2.46(s, CH₃₋₂), 7.80(s,

H-5), 2.30(s-CH₃₋₁) 3.50(m, H-1'), 6.08(m-H-2), 5.30(dd, H-3', J=10, 1H-2), 5.09(dd, H-3', J=16, 1H-2). **5a** δ 1.50 (d, CH₃₋₂, J=6H-2), 2.05(s, CH₃₋₃), 2.45(s, CH₃₋₂), 2.90(dd, H-3', J=12, 8H-2), 3.42(dd, H-3', J=16, 6H-2), 5.10(m, H-2'), 8.20(d, H-8, J=10H-2), 6.75(d, H-9, J=10 H-2). **5b** δ 1.62(d-CH₃₋₂, J=6H-2), 2.05 (s, CH₃₋₃), 2.42(s, CH₃₋₂), 3.02(dd, H-3', J=12, 8H-2), 3.56(dd, H-3', J=12, 6H-2), 5.25(m, H-2'), 8.12(s-H-5). **5c** δ 1.62 (d, CH₃₋₂', J=6H-2), 2.07(s, CH₃₋₃), 2.41(s, CH₃₋₂), 5.18(m, H-2'), 3.00(dd, H-3', J=12, 8H-2), 3.56(dd, H-3', J=12, 6H-2), 8.06(s, H-5). **5d** δ 1.61(d, CH₃₋₂, J=6H-2), 2.06 (s, CH₃₋₃), 2.44(s, CH₃₋₂), 2.30(s, CH₃₋₈), 3.02(dd, H-3', J=12, 8H-2), 3.52(dd, H-3', J=12, 6H-2), 5.20(m, H-2'), 8.20(s, H-8) [3].

Analytical experimental

The melting points were determined in open capillary tubes and are uncorrected. Elemental analysis was determined using Carlo Erba model 1108. IR Spectras were recorded on Perkin-Elmer 881 spectrophotometer in KBr(cm⁻¹) and ¹H NMR spectra on a Bruker advanced DXP 200 spectrophotometer using TMS as internal reference [4].

Anti-inflammatory activity

Animals

Adult male albino rats (wistar strain) weighing between 150-250g were used for anti-inflammatory study. They were fed on commercial diet and water *ad libitum*. All the animals were acclimatized for a week before use. The room temperature was maintained at 25±1°C. The experimental protocol was approved by institutional animal ethical committee [5].

Experimental design for anti-inflammatory activity

The anti-inflammatory activity of the synthesized compounds was carried out by the method of Winter et al. Wistar rats weighing 150-250 g were used for the edema test. Animals were divided into six rats per group. Rats were put on fast for 18 h. prior to the experiment. The standard drug ibuprofen (100mg / kg body wt.) and the drug (100mg / kg body wt.) were given orally as a suspension in 0.1% sodium carboxy methylcellulose as vehicle. One h later 0.1 ml of carageenin solution in saline was injected in the subplantar region of the right hind paw of each rats. After 3 h of carageenin injection the reduction in paw volume compared to control vehicle was measured using plethysmometer. (Table-1) [6].

Table 1: Characterization and antiinflammatory activity of synthesized compounds

Compounds	M.P(°C)	Yield (%)	Paw Volume(% protection)
Control			2.25± 0.08
Ibuprofen			0.89± 0.13(60)
4a	134	34	1.80± 0.19(20)
4b	147	36	1.73± 0.14(23)
4c	132	57	0.77± 0.18(66)
4d	176	74	1.08± 0.02(52)
5a	145	45	1.40± 0.01(38)
5b	139	51	1.64± 0.11(27)

Statistical analysis

Values for anti-inflammatory activity were expressed as "mean increase in paw volume ±SEM". The significance of difference between means was determined by student's t-test values of p<0.05 were considered significant and p<0.01 as highly significant.

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